



**Shoubra faculty
of Engineering**

2nd year Mech. 2017/2018



Electronics Engineering

Lecture 1

Introduction to Electric Circuits

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Feb. 2018

Electronics in Hand



Course Information

Title	Electronics Engineering
Lecturer	Dr. Sawsan Abdellatif
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Assessment (100)	<ul style="list-style-type: none">▪ Final Exam (70)▪ Course work (30)<ul style="list-style-type: none">• Midterm (12)• Section work (10)• Quiz/assignments (8)

Course Schedule

	Topics	(# Lectures)
Part (1)	Introduction to Electric Circuits	1
	Ohms law, Energy and Power	1
	Series and Parallel Circuits	1
	Series-Parallel Circuits	1
	Circuit Theorems and Conversions	1
	Introduction to AC current and Voltage	1
	RC, RL, RLC circuits	1
	Midterm Exam	
Part (2)	Introduction to Electronics	1
	Diode and its applications	1
	Bipolar Junction Transistor and its applications	1
	Logic Gates	1
	Quiz and course closeout	1

References

- “Principles of Electric Circuits-Conventional Current Version”,
9th Edition, Floyd
- “Electronic Devices”, 9th Edition, Floyd
- “Digital Fundamentals”, 9th Edition, Floyd



Quantities and Units

Electrical Quantities

QUANTITY	SYMBOL	SI UNIT	SYMBOL
Capacitance	C	Farad	F
Charge	Q	Coulomb	C
Conductance	G	Siemens	S
Energy (work)	W	Joule	J
Frequency	f	Hertz	Hz
Impedance	Z	Ohm	Ω
Inductance	L	Henry	H
Power	P	Watt	W
Reactance	X	Ohm	Ω
Resistance	R	Ohm	Ω
Voltage	V	Volt	V

Metric Prefixes

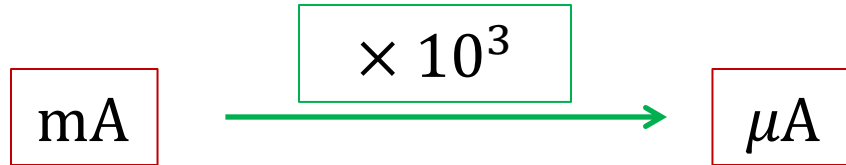
METRIC PREFIX	SYMBOL	POWER OF TEN	VALUE
femto	f	10^{-15}	one-quadrillionth
pico	p	10^{-12}	one-trillionth
nano	n	10^{-9}	one-billionth
micro	μ	10^{-6}	one-millionth
milli	m	10^{-3}	one-thousandth
kilo	k	10^3	one thousand
mega	M	10^6	one million
giga	G	10^9	one billion
tera	T	10^{12}	one trillion

Metric Unit Conversions

- How to Convert from one unit with a metric prefix to another

Example

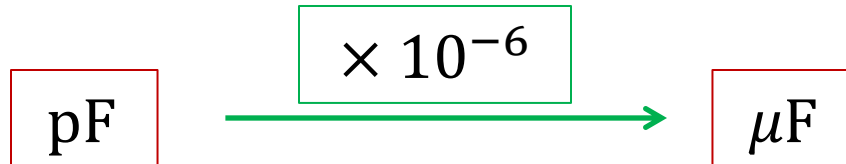
Convert 0.15 milliamperes (0.15 mA) to microamperes (μA).



- $0.15 \text{ mA} = 0.15 \times 10^3 \mu A = 150 \mu A$

Example

Convert 47,000 picofarads (47,000 pF) to microfarads (μF).



- $47000 \text{ pF} = 47000 \times 10^{-6} \mu F = 0.047 \mu F$

Metric Prefix: 10^x

$\times 10^{(x-y)}$

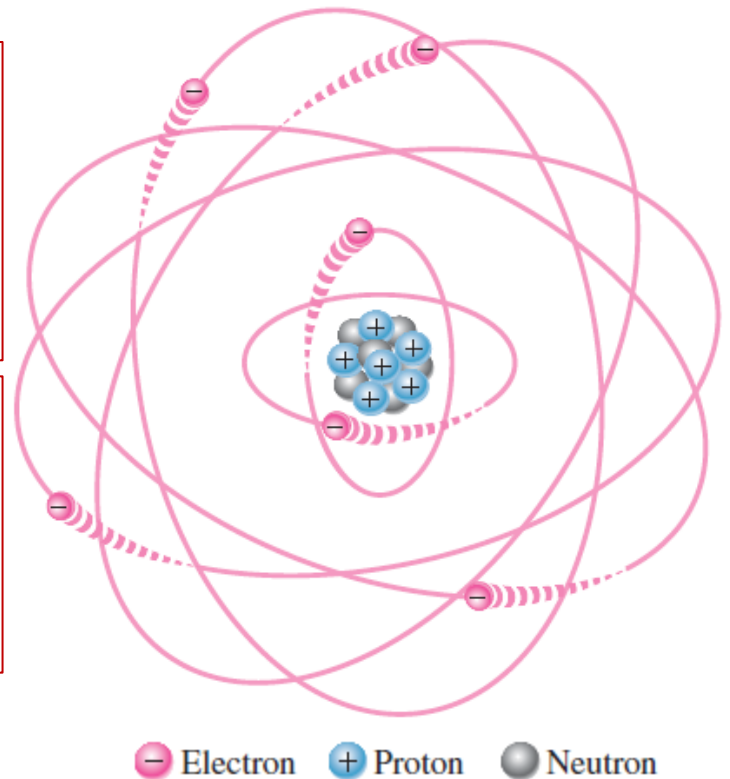
Metric Prefix: 10^y



Voltage, Current, and Resistance

Atomic Structure

- An **atom** is the smallest particle of an element that retains its characteristics.
- An **atom** consists of central **nucleus** surrounded by orbiting **electrons**.
- The **nucleus** consists of **positively** charged particles called **protons** and **uncharged** particles called **neutrons**.
- The basic particles of **negative** charge are called **electrons**, which orbit the nucleus.



Atomic Structure (cont'd)

Atomic Number

- The atomic number equals the **number of protons** in the nucleus.
- For example, hydrogen has an atomic number of 1.

- In the **normal** (or **neutral**) state, all atoms of a given element have the **same number of electrons as protons**; the positive charges cancel the negative charges, and the atom has a **net charge of zero**.

Shells, Orbits, and Energy Levels

- Electrons **orbit** the nucleus of an atom at certain distances from the nucleus called **Orbits**.
- Each orbit corresponds to a different **energy level** called shell
- Maximum **number** of **electrons** in each shell = $2N^2$, $N=1,2,3..$ is number of shell (i.e., Max. number of electrons 2, 8, 18, 32,)

Valance Electrons

- Electrons in the **outermost** shell of an atom
- Have **highest energy** levels
- Relatively **loosely bound** to the atom
- Contribute to **chemical reactions**
- Determine the material's **electrical properties**

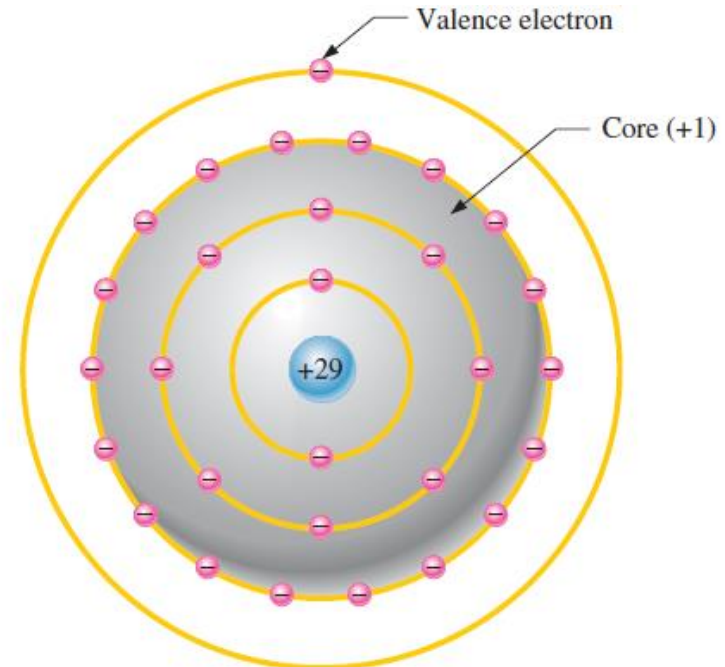


Fig. The copper atom

The Copper Atom

- Atomic number of Cu=29
- The outermost shell (the valence shell) has only **one valence electron**.
- When the valence electron gains sufficient thermal energy, it can leave its atom and become a **free electron**.

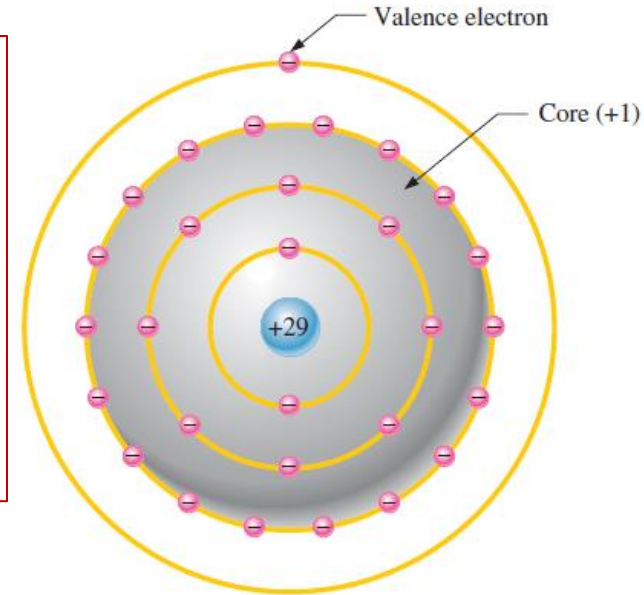


Fig. The copper atom

- In a piece of copper **at room temperature**, a “sea” of free electrons is present.
- These electrons are **not bound to a given atom** but are free to move in the copper material.
- Free electrons make copper an **excellent conductor** and make electrical current possible.

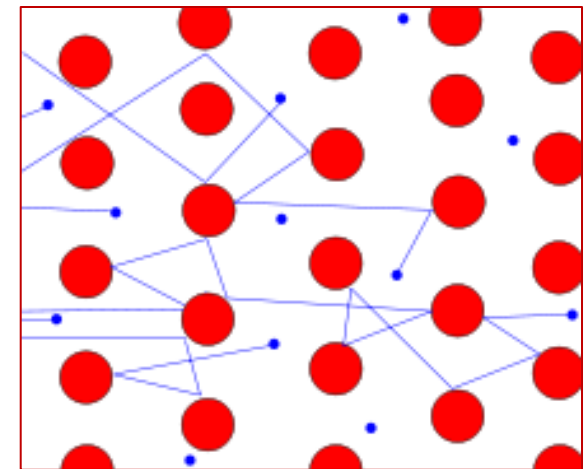


Fig. Copper Crystal

Electrical Charge

- Electrical charge (Q) is an electrical property of matter that exists because of an **excess or deficiency of electrons**.
- *Static electricity is the presence of a net positive or negative charge in a material.*
- Charges of **opposite polarity** are **attracted (attraction force)** to each other, and charges of the **same polarity** are **repelled (repulsion force)**.
- This force, called an electric field.



(a) Uncharged:
no force



(b) Opposite
charges
attract



(c) Like positive
charges repel



(d) Like negative
charges repel

Coulomb: The Unit of Charge

- Electrical charge (Q) is measured in coulombs (C).
- One coulomb is the total charge of 6.25×10^{18} electrons.
- A single electron has a charge $e = 1.6 \times 10^{-19} \text{ C}$

Example

How many coulombs do 93.8×10^{16} electrons represent?

$$Q = \text{Number of Electrons} \times e = 93.8 \times 10^{16} \times 1.6 \times 10^{-19} = 0.15 \text{ C}$$

Example

How many electrons does it take to have 3 C of charge?

The Voltage

- The voltage is defined as the **work** done per unit charge **to move a charge** between two points.
- *The voltage is also known as **potential difference** or **electromotive force (emf)**.*

$$V = \frac{W}{Q}$$

where V is voltage in volts (V), W is energy in joules (J), and Q is charge in coulombs (C).

Example

If 50 J of energy are required to move 10 C of charge, what is the voltage?

$$V = \frac{W}{Q} = \frac{50 \text{ J}}{10 \text{ C}} = 5 \text{ V}$$

The Current

- Voltage provides energy to electrons, allowing them to move through a circuit. This **movement of electrons** produces current.

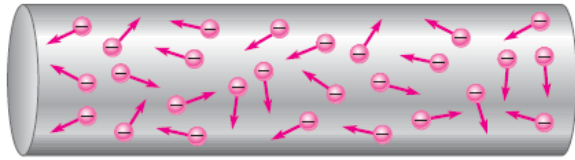


Fig. Random motion of free electrons in a material. (**when no voltage applied**)

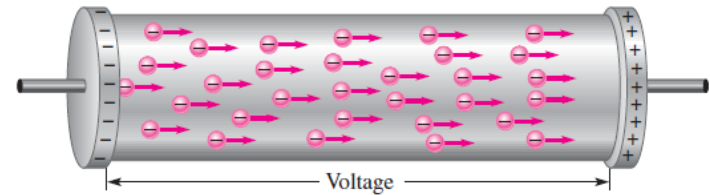
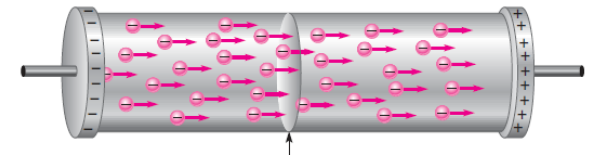


Fig. Electrons flow from -ve to +ve **when a voltage is applied** across a conductive material.

- Electrical current (I) is the **rate of flow of charge**.

$$I = \frac{Q}{t}$$

I is current in amperes (A), Q is charge in coulombs (C), and t is time in seconds (s).



Example Ten coulombs of charge flow past a given point in a wire in 2 s. What is the current in amperes?

$$I = \frac{Q}{t} = \frac{10 \text{ C}}{2 \text{ s}} = 5 \text{ A}$$

The Voltage Source and Current Source

- An ideal **voltage source** can provide a **constant voltage** for any load.
- An ideal **current source** can provide a **constant current** for any load.

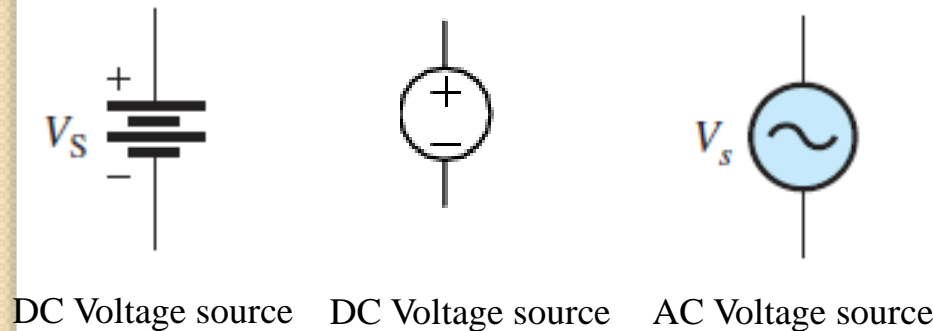


Fig. Symbols of **Voltage source**

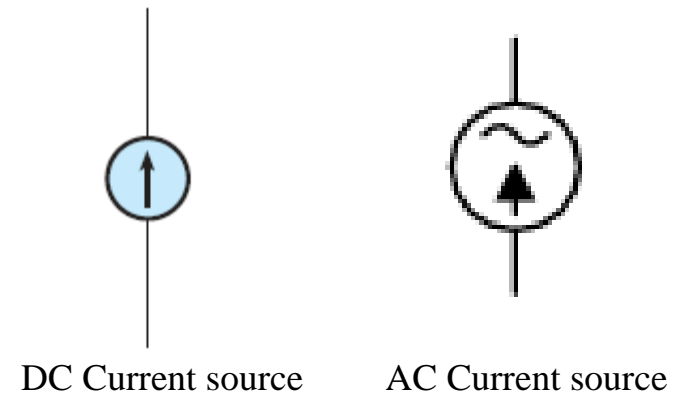


Fig. Symbols of **Current source**

- *Batteries, Solar Cells, DC generator, power supply... are examples of DC voltage source*

Resistance and Conductance

- Resistance (R) is the **opposition** to current.
- The unit of Resistance is **ohms** (Ω)

- Conductance (G) is the **reciprocal** of resistance.
- The unit of conductance is the **Siemens** (S)

$$G = \frac{1}{R}$$



Fig. Symbol of resistance



Fig. Symbol of conductance

The Resistors

- A **component** that is designed to have a **certain** amount of **resistance** is called a resistor.
- Resistors are use to **limit current** in a circuit, to **divide voltage**.

Fixed Resistors

Their resistance values are **fixed**

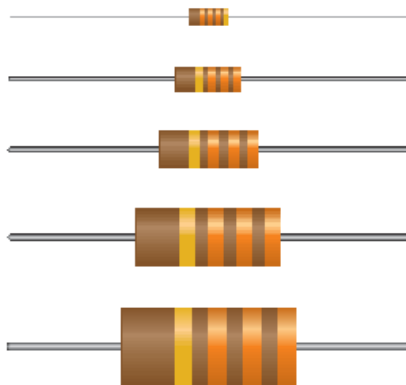
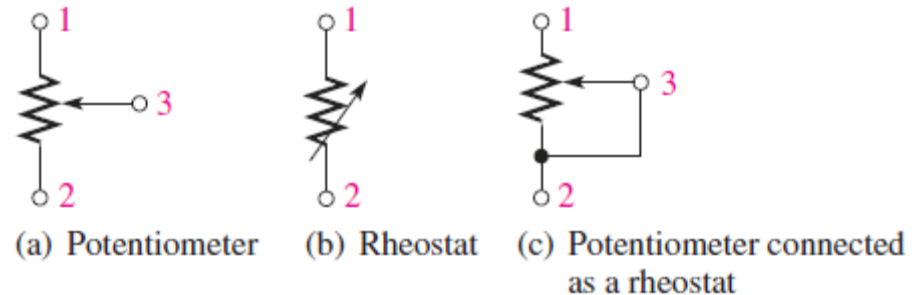


Fig. Carbon-composition resistors

Variable Resistors

Their resistance values can be **changed** easily.



Potentiometer	3 Terminals	Divide voltage
Rheostat	2 Terminals	Control current

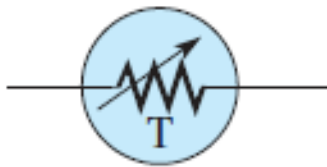
Resistors (Cont'd)

Variable Resistance sensors

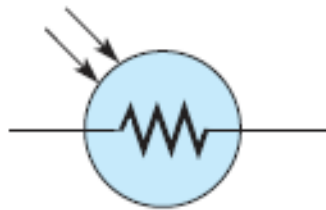
- Many sensors operate on the **concept of a variable resistance**, in which a **physical quantity alters the electrical resistance**.

Examples:

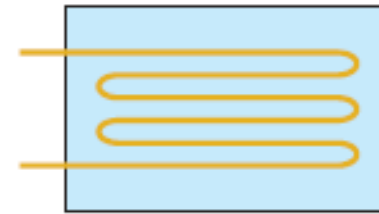
- **Thermistors**: the resistance is a function of **temperature**
- **Photoconductive cells**: the resistance is a function of **light**
- **Strain gauges**: the resistance is function of **force** applied



(a) Thermistor



(b) Photoconductive cell



(c) Strain gauge

The Electric Circuit

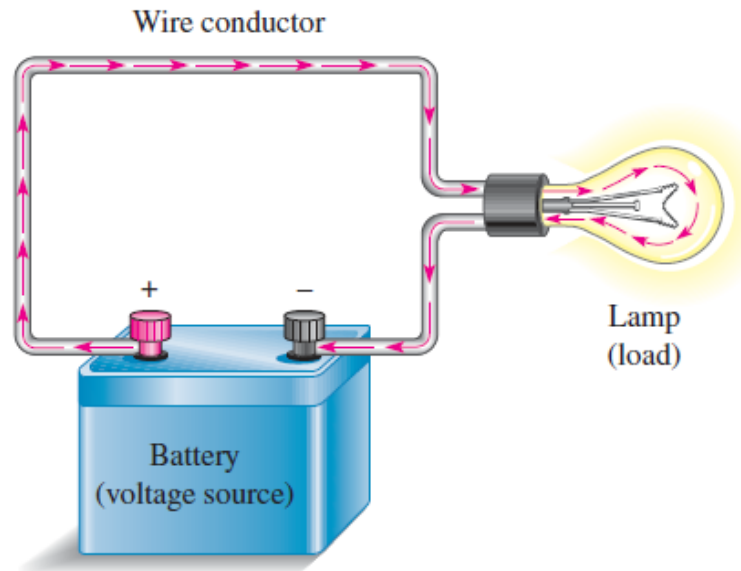


Fig. (a) Simple electric circuit



Fig. Schematic for the circuit in (a)

The Electric Circuit (cont'd)

Closed Circuit

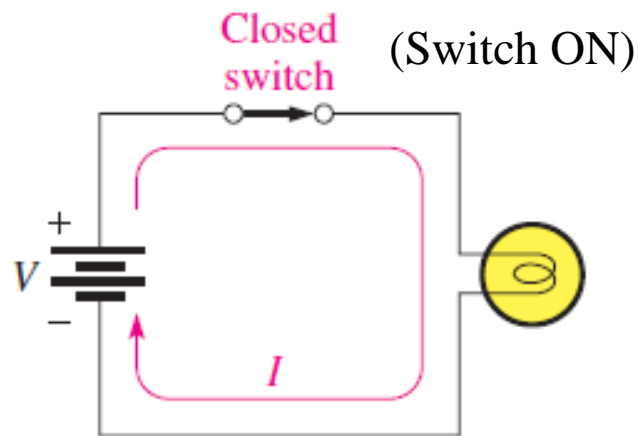


Fig. There is **current I** because there is a complete current path.

Open Circuit

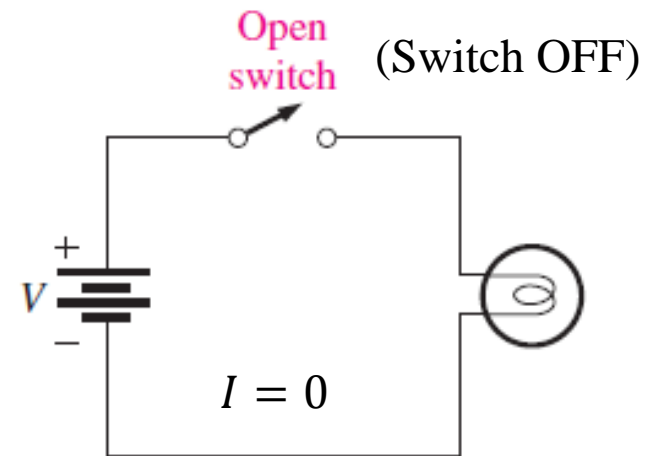
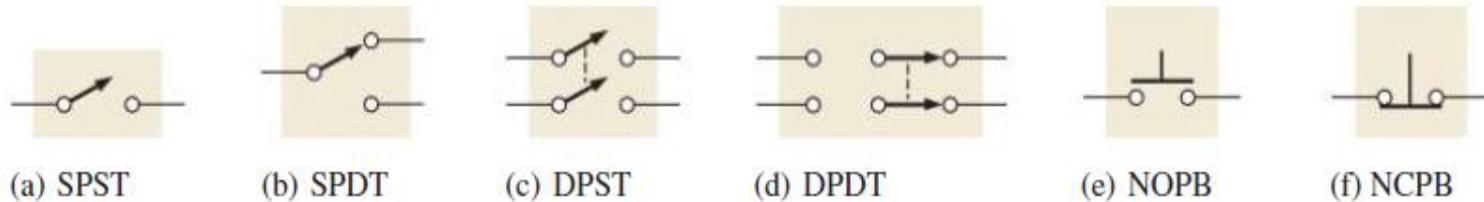


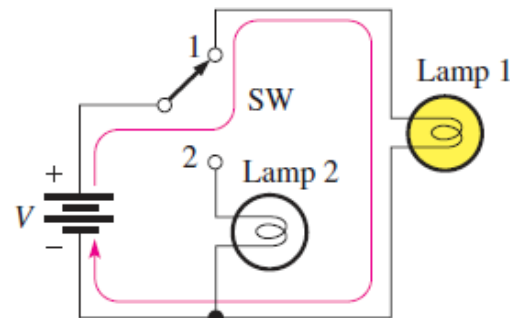
Fig. There is **no current** because the path is broken.

The Electric Circuit (cont'd)

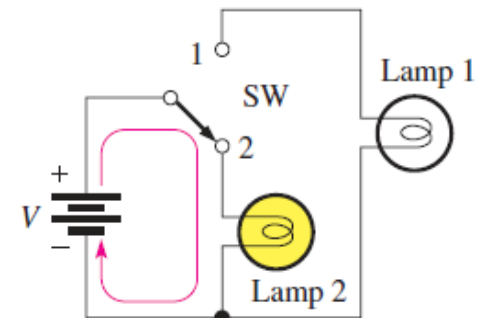
Types of Switches



- **SPST:** Single Pole Single Through
- **SPDT:** Single Pole Double Through
- **DPST:** Double Pole Single Through
- **DPDT:** Double Pole Double Through
- **NOPB:** Normally Opened Push Button
- **NCPB:** Normally Closed Push Button



(b) A schematic showing Lamp 1 on and Lamp 2 off



(c) A schematic showing Lamp 2 on and Lamp 1 off

Fig. An example of an SPDT switch controlling two lamps.

The Electric Circuit (cont'd)

Protective Devices

- Fuses and circuit breakers are used to create an open circuit when the current exceeds a certain value due to an abnormal condition in a circuit.



(d) Fuse symbol

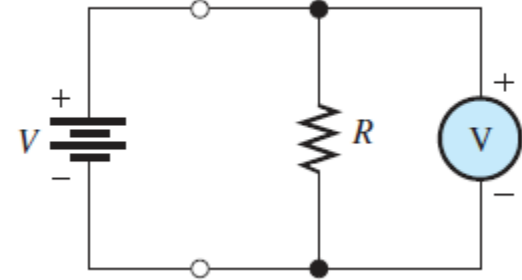


(e) Circuit breaker symbol

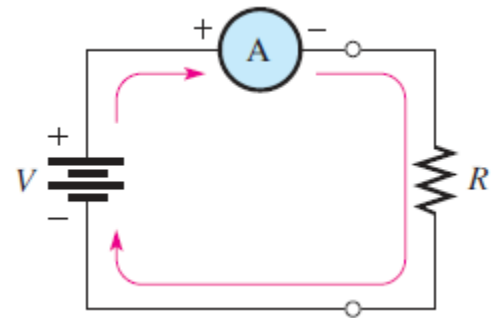
- When a fuse is “blown,” it must be replaced; but when a circuit breaker opens, it can be reset and reused repeatedly.

Measuring Instruments

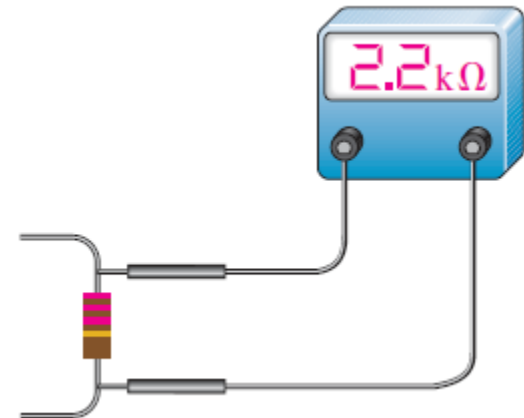
Voltmeter: Measure Voltage



Ammeter: Measure Current



Ohmmeter: Measure Resistance





Thanks for attention